

# An Analysis of the Prospects of the Use of Magnetic Water Treatment in Foundry Engineering

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## Abstract

Scientists are currently focused on creating technologies that produce positive results without affecting the environment. One such technology is magnetic water treatment. In this paper, an analytical review of publications devoted to the application of magnetic treatment of water in various branches of engineering, agriculture, and medicine is carried out. Current views on the structure of water molecules, as well as the theories explaining the influence of the magnetic treatment of water on its properties, are reviewed. The results of studies of the influence of water treated by a magnetic field on the properties of molding sand are analyzed, including those in which the authors of the article took part. It is shown that the magnetic treatment of still water can increase the green strength of the molding sand containing this water from 0.035 to 0.052 MPa, and that of water in motion to 0.075 MPa. Thanks to this, the amount of binder in the molding sand can be reduced. It is concluded that the use of magnetically treated water in foundries is promising.

## Keywords:

water, magnetic field, magnetic water treatment, molding sand, green strength

## 1. INTRODUCTION

G. Piccardi suggested in the 1930's that there was an interrelation between the geomagnetic field and water properties [1]. However, the first real interest among specialists in magnetic water treatment arose after 1945, when the Belgian engineer T. Vermeiren patented a simple and effective way to combat salt deposits in steam boilers [2]. Water containing hardness salts was passed through a magnetic field. In a fairly short period, studies on the effects of magnetic and electromagnetic water treatment in various industries were conducted (Fig. 1):

- in the power industry to reduce the formation of scales in heat exchangers [3–11];
- in the production of building materials [12–17];
- in agriculture to increase the yield of various crops [18–20];
- in the neutralization of wastewater [21];
- in medicine and biology [22];
- in foundry engineering for the preparation of molding sands [22–28].

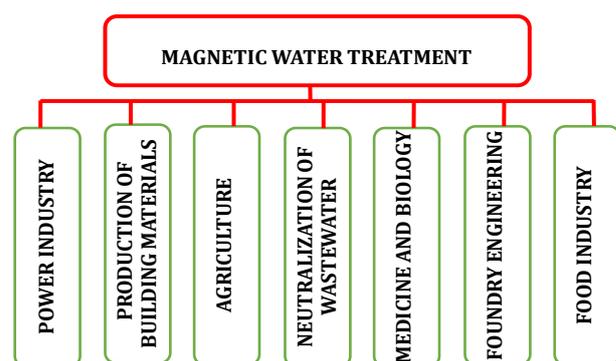


Fig. 1. Magnetic water treatment applications

In the majority of cases, the authors noted a positive effect of magnetic or electromagnetic water treatment. However, V. Ochkov [29] was skeptical about the data available in the literature. He explained the positive results of magnetic water treatment as a) accidental; b) a manifestation of the “human factor”; c) a marketing campaign by the manufacturers of special equipment.

The present review focuses on the analysis of the positive effect of electromagnetic water treatment on the properties of the molding sands in which it is used. Theories that have appeared in recent years are used to explain the available results.

## 2. STRUCTURE OF WATER AND HYPOTHESES EXPLAINING CHANGES IN ITS PROPERTIES UNDER MAGNETIC TREATMENT

In 1933, D.J. Bernal and R.H. Fowler [30] created the classical theory of water structure. They first showed that each water molecule is surrounded by four others. Due to the existence of directional intermolecular bonds, the arrangement of molecules resembles tetrahedral coordination.

W. Ramsay and J. Shields [31], W. Sutherland [32], G.W. Stewart [33], R. Mecke [34], M. Yel'yashevich [35], M. Magat [36], N. Bjerrum [37], L. Pauling [38], J. Lennard-Jones and J. Pople [39] made great contributions to the development of water structure theory. In recent years, the fundamental works of X.F. Pang [40, 41] have been published.

The predominant opinion is that "most hydrogen-bonded chains of water molecules can mutually unite and form some closed configurations by connecting the head and tail of the linear chains with hydrogen bonds. These closed configurations can include chains with hydrogen bonds containing 2 (dimer), 3, 4, 5, 6, or more water molecules" [40]. This view can be used to explain the magnetizability of water.

Numerous experiments have shown that water can be magnetized under the influence of a magnetic field, although the magnetization effect is small. Based on the analysis of a large amount of experimental data, the authors [21, 22, 41–44] showed that this changes its optical and electromagnetic properties, surface tension force, dielectric permittivity, viscosity, crystallization and boiling temperatures. For example, the authors [42] indicate that the surface tension of water after magnetic treatment decreased from 72.44 mN/m to 57.62 mN/m (1 T intensity field with treatment time for 13 min at 25°C). The authors [43] point out that the refractive index of magnetized water was increased by 0.1% from 1.333 to 1.335 under the influence of a magnetic field (10 T at 25°C). The authors [43] also observed a slightly increasing melting point (5.6 mK) of water after magnetic treatment as well as an increase in water's vaporization enthalpy after magnetic treatment (45–65 mT, 22°C) from 58.86 to 68.86 kJ/mol. The studies [41] also showed significant differences in infrared absorption spectra and Raman spectra of magnetized and ordinary water. Raman spectra peaks for the magnetized water are larger than that without magnetized water (the positions of these peaks do not change). The strengths of the 1500 cm<sup>-1</sup> peak in the magnetized and pure waters are 148 and 124, respectively. The authors [41] explain this result by the intensification of "ring proton currents" in the closed chains of water molecules.

Hypotheses explaining the essence of the magnetic field effect on water and aqueous solutions can be divided into three groups.

The first group of hypotheses proceeds from the fact that water always contains impurities. According to the theories

of this group, the spontaneous formation and decay of colloidal complexes of metal cations: Ca<sup>2+</sup>, Mg<sup>2+</sup>, Fe<sup>2+</sup> and Fe<sup>3+</sup>, occurs under the influence of a magnetic field in the treated water and fragments of their deca further form the centers of nucleation of inorganic salts. In the presence of Fe<sup>3+</sup> cations and the smallest ferromagnetic Fe<sub>2</sub>O<sub>3</sub> particles in water, the formation "of colloidal hydrophobic sols of Fe<sup>3+</sup> cations with chlorine Cl<sup>-</sup> anions and neutral H<sub>2</sub>O molecules having the general formula [xFe<sub>2</sub>O<sub>3</sub>·yH<sub>2</sub>O·zFe<sup>3+</sup>]<sup>+</sup>·3zCl<sup>-</sup>, which may cause the formation of nucleation centers whose surface adsorbs calcium Ca<sup>2+</sup> and magnesium Mg<sup>2+</sup> cations (forming the basis of the carbonate hardness of water)" [45]. It should be noted that the theories of this group satisfactorily explain the positive effect of magnetic treatment of water to prevent or reduce scale formation in pipes and heat exchangers.

Hypotheses of the second group explain the effect of a magnetic field on water by means of the polarization of dissolved ions and the deformation of their hydration shells under the action of the magnetic field [46]. In addition, it is assumed that the effect of the magnetic field on the Ca<sup>2+</sup>, Mg<sup>2+</sup>, Fe<sup>2+</sup>, and Fe<sup>3+</sup> ions dissolved in water may also be associated with the generation of a weak electric current in the moving water flow or with pressure pulsation [47].

Hypotheses of the third group postulate that the magnetic field directly affects the structure of water associates due to the dipole polarization of water molecules, which are formed from many H<sub>2</sub>O molecules bonded to each other by low-energy intermolecular Van der Waals forces, dipole-dipole interactions and hydrogen bonds, which can cause the deformation of hydrogen bonds and their partial breakage, and the migration of mobile protons H<sup>+</sup> within associative elements of water and uniting of H<sub>2</sub>O molecules into temporary associates – clusters [40, 48].

The above hypotheses do not fully cover all of the assumptions and views on the essence of phenomena occurring during the magnetic treatment of water. At the same time, a large number of experiments confirming changes occurring in water under the influence of a magnetic field eliminates doubts as to the validity of the observed phenomena.

## 3. THE USE OF MAGNETIC WATER TREATMENT IN FOUNDRY ENGINEERING

An essential reserve for improving the quality of casting into sand molds is to improve the properties of molding materials used for their production. An important point in this is the use of environmentally friendly technologies and materials.

The strength of green molding sands depends mainly on the properties (the adhesive properties and the contact angle) of the liquid and semi-liquid films covering the sand grains (water, moistened clay, etc.). Thus, by influencing these films, it is possible to influence the strength properties of the molding sands as a whole.

Boldin et al. [49] explain the formation of the strength properties of green molding sand by the polarity of the water molecule and the presence of an electric double electric charge on the micelles of the clay binder.

The authors [49–52] explain the formation of the strength of green molding sand by the fact that on the surface of solid

particles (sand and clay), a double electric layer is formed by hydrogen ions and hydroxyl groups. There is a spontaneous, unipolar orientation of dipoles and liquid ions on the surface of sand and clay. The degree of orientation of the particles decreases with distance from the surface. At the same time, according to Boldin et al. [49], the disoriented particles have a “wedging” effect and prevent particle convergence.

It can be assumed that magnetically treated water will have dipoles with preferential orientation with respect to the surface of mineral particles (sand and clay). This will decrease the “wedging” effect and provide a better particle approach and, consequently, the greater strength of the sand.

In a number of studies [22–28], water, clay slurries, aqueous solutions of sols included in the molding sand, and the green molding sand were subjected to magnetic treatment. In addition, different starting materials were used, from which different compositions of molding sand were prepared. The action was carried out both by electromagnets and permanent magnets, and the mixing time of the components was different. Therefore, it is impossible to compare the quantitative indicators of the effect of magnetic treatment.

Let us analyze the results of individual works separately in order to identify general patterns.

V. Klassen [23] supplies data on the increase in the compressive strength of the green molding sand mixture (93.2% quartz sand, 4.8% clay, and 2.9% water) 2 times after magnetic treatment of water with a simultaneous increase in gas permeability.

In [27] the magnetic treatment of water allowed the strength of the sand-cement molding sand to be increased by 30%.

Yu. Vasin et al. [53] also conducted studies with sand and bentonite mixtures (100 wt. parts of quartz sand, 10 wt. parts of bentonite and 4.5 wt. parts of water). The application of magnetized water increased the strength from 40 to 52 kPa (1.3 times) (Fig. 2) and gas permeability from 287 to 313 units (1.44 times). It is remarkable that the effect of the magnetic treatment of water was retained for more than 1 hour.

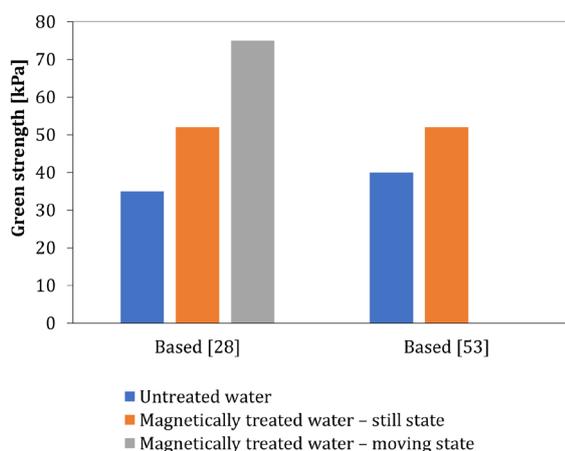


Fig. 2. Green strength of molding sands

The authors of the research [24–27] focused their attention on the study of the effect of movement velocity of sodium silicate, clay slurry, and an aqueous NaOH solution on

the strength of different molding sands. The magnetic treatment of still sodium silicate made it possible to increase the strength of the molding sand (97 wt. parts of quartz sand, 3 wt. parts of molding clay, 1 wt. part of NaOH, 5 wt. parts of sodium silicate with a density of 1420 kg/m<sup>3</sup>) by 1.6 times. The movement of sodium silicate through the electromagnet at a velocity of 1 g/s additionally increased the strength of the sand by 10% compared to the fixed sodium silicate treatment. It was found that the positive results of the magnetic treatment was retained after even 9 hours [25]. Moving 0.5% aqueous NaOH solution through the electromagnetic treatment apparatus at a speed of 0.6 m/s ensured an increase the strength of the molding sand (90% quartz sand, 1% clay and 3.5% aqueous NaOH solution) 1.35 times compared to the fixed solution treatment [24].

L. Dan et al. [28] conducted research on the effect of various factors on the green strength of the molding sand, which contained quartz sand (76.2%), molding clay from the Chasiv Yar deposit in the form of powder (19%) and water (4.8%). During the experiments, water was treated by electromagnet at still or by stirring in the vessel; the treatment time and the electric power supplied to the electromagnet were varied.

The study [28] showed that even the treatment of still water with an alternating magnetic field increased the green strength of the molding sand containing this water from 0.035 MPa to 0.052 MPa compared to the molding sand that contained untreated water (1.48 times). Stirring with a glass stirrer at ~1 revolution per second increased the green strength to 0.075 MPa (an additional 1.44 times) (Fig. 2).

To achieve maximum results for the green strength of the molding sand, it was sufficient to conduct magnetic water treatment for 2.5 minutes. As in the studies [25, 53], the effect of the magnetic treatment of water was maintained for a long period of time. Even after 120 hours, the green strength of the samples made from sand containing magnetically treated water was 1.3 times higher than that of the control samples.

Analysis of the use of magnetically treated water in foundry engineering has shown the following positive results: a) magnetic treatment of water and water binders indisputably increases the green strength of molding sand and improves its gas permeability; b) magnetic treatment of moving water is more effective compared to the magnetic treatment of still water; c) the positive effect of magnetic treatment of water on the properties of molding sand is maintained for a long time. At the same time, there is no consensus in the literature about the modes of magnetic water treatment required to achieve the best results.

#### 4. CONCLUSIONS

Analysis of the literature on the magnetic treatment of water has shown the following:

- water changes its properties under the influence of a magnetic field: electro-physical, optical, surface tension, temperatures of phase transformations, infrared absorption spectra, Raman spectra, etc.;
- to date, there is no unified theory explaining changes in water properties under the action of a magnetic field;

- the application of water treated by a magnetic field to reduce scale formation in heat exchangers and pipes, in agriculture, in the manufacture of concrete products, in biology and medicine provides a positive effect;
- the use of water magnetization and aqueous binders in molding sands increases their green strength. Magnetic treatment of still water can increase the green strength of the molding sand containing this water from 0.035 to 0.052 MPa, and that of water in motion to 0.075 MPa. It can reduce the amount of binder used in the molding mixtures;
- long-term retention of improved properties of the molding sands opens up the prospect of using the magnetic treatment of water and aqueous binders in the industry;
- to make a final decision on the use of magnetic water treatment in foundry engineering, it is necessary to carry out comprehensive research aimed at establishing the optimal techniques and modes of such treatment.

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### REFERENCES

- [1] Piccardi G. (1962). *The Chemical Basis Medical Climatology*. Springfield: Thomas.
- [2] Vermeiren T.I.S. (1957). Patent France No. 1145070 C02. Lille, Ministère de l'Industrie et du Commerce, Service de la Propriété Industrielle.
- [3] Martynova O.I., Kopylov A.S. & Ochkov V.F. (1978). Mechanism and scale formation control in MSF (Multi Stage Flash Desalination) plant using an electromagnetic apparatus. *Proceedings of 6<sup>th</sup> International Symposium Flash Water from the Sea, September 17–22. Madrid*, 240–251 [s.n.]. [http://twtmpei.ac.ru/ochkov/Grand\\_Canaria/index.htm](http://twtmpei.ac.ru/ochkov/Grand_Canaria/index.htm) [accessed: 21.03.2023].
- [4] Martynova O.I., Kopylov A.S., Tebenikhin Ye.F. & Ochkov V.F. (1979). K mekhanizmu vliyaniya magnitnoy obrabotki vody na protsessy nakipeobrazovaniya i korrozii. *Teploenergetika*, 26(6), 67–69 [Мартынова О.И., Копылов А.С., Тебенихин Е.Ф. & Чков В.Ф. (1979). К механизму влияния магнитной обработки воды на процессы накипеобразования и коррозии, *Теплоэнергетика*, 26(6), 67–69. [http://twtmpei.ac.ru/ochkov/MO/art\\_4\\_M\\_K\\_T\\_O.htm](http://twtmpei.ac.ru/ochkov/MO/art_4_M_K_T_O.htm) [accessed: 21.03.2023].
- [5] Kozic V., Hamler A., Ban I. & Lipus L.C. (2010). Magnetic water treatment for scale control in heating and alkaline conditions. *Desalination and Water Treatment*, 22(1–3), 65–71. Doi: <https://doi.org/10.5004/dwt.2010.1549>.
- [6] Gabrielli C., Jaouhari R., Maurin G. & Keddad M. (2001). Magnetic water treatment for scale prevention. *Water Research*, 35(13), 3249–3259. Doi: [https://doi.org/10.1016/S0043-1354\(01\)00010-0](https://doi.org/10.1016/S0043-1354(01)00010-0).
- [7] Lin L., Jiang W., Xu X. & Xu P. (2020). A critical review of the application of electromagnetic fields for scaling control in water systems: mechanisms, characterization, and operation. *Clean Water*, 25, 1–19. <https://www.nature.com/articles/s41545-020-0071-9> [accessed: 21.03.2023].
- [8] Mosin O. & Ignatov I. (2015). Practical implementation of magnetic water treatment to eliminate scaling salts. *Journal of Health, Medicine and Nursing*, 10, 111–125. <https://iuste.org/Journals/index.php/IHMN/article/view/19684>.
- [9] Dalas E. & Koutsoukos P.G. (1989). The effect of magnetic fields on calcium carbonate scale formation. *Journal of Crystal Growth*, 96(4), 802–806. Doi: [https://doi.org/10.1016/0022-0248\(89\)90640-4](https://doi.org/10.1016/0022-0248(89)90640-4).
- [10] Alabi A., Chiesa M., Garlisi C. & Palmisano G. (2015). Advances in anti-scale magnetic water treatment. *Environmental Science: Water Research & Technology*, 4, 408–425. Doi: <https://doi.org/10.1039/C5EW00052A>.
- [11] Baker J.S. & Judd S.J. (1996). Magnetic amelioration of scale formation. *Water Research*, 30(2), 247–260. Doi: [https://doi.org/10.1016/0043-1354\(95\)00184-0](https://doi.org/10.1016/0043-1354(95)00184-0).
- [12] Faris A.S., Al-Mahaidi R. & Jadooe A. (2014). Implementation of magnetized water to improve the properties of concrete. *International Journal of Civil Engineering and Technology*, 5(10), 43–57. <http://hdl.handle.net/1959.3/391689>.
- [13] Narmatha M., Arulraj P. & Bari J.A. (2021). Effect of magnetic water treatment for mixing and curing on structural concrete. *Materials Today: Proceedings*, 37(2), 671–676. Doi: <https://doi.org/10.1016/j.matpr.2020.05.633>.
- [14] Su N., Wu Y.-H. & Mar C.-Y. (2000). Effect of magnetic water on the engineering properties of concrete granulated blast-furnace slag. *Cement and Concrete Research*, 30(4), 599–605. Doi: [https://doi.org/10.1016/S0008-8846\(00\)00215-5](https://doi.org/10.1016/S0008-8846(00)00215-5).
- [15] Malathy R., Narayanan K. & Mayakrishnan P. (2022). Performance of prestressed concrete beams using magnetic water for concrete mixing. *Journal of Adhesion Science and Technology*, 36(6), 666–684. Doi: <https://doi.org/10.1080/01694243.2021.1936383>.
- [16] Ramalingam M., Narayanan K., Masilamani A., Kathirvel P., Murali G. & Vatin N.I. (2022). Influence of magnetic water on concrete properties with different magnetic field exposure times. *Materials*, 15(12), 4291. Doi: <https://doi.org/10.3390/ma15124291>.
- [17] Malathy R., Narayanan K. & Baranidharan S. (2017). Effect of magnetic water on mixing and curing of M25 grade concrete. *International Journal of Chemical Technology Research*, 10(11), 131–139.
- [18] Yusuf K.O., Tokosi R.O. & Raji M. (2022). Effect of magnetic treatment of irrigation water on germination, growth, yield and popping-quality of popcorn under deficit irrigation. *Notulae Scientia Biologicae*, 14(3), 11323. Doi: <https://doi.org/10.55779/nsb14311323>.
- [19] Bogatin J., Bondarenko N.P., Gak E.Z., Rokhinson E.E. & Ananyev I.P. (1999). Magnetic treatment of irrigation water: Experimental results and application conditions. *Environmental Science Technology*, 33(8), 1280–1285. Doi: <https://doi.org/10.1021/es980172k>.
- [20] Maheshwari B.L. & Grewal H.S. (2009). Magnetic treatment of irrigation water: Its effects on vegetable crop yield and water productivity. *Journal of Agricultural Water Management*, 96(8), 1229–1236. Doi: <https://doi.org/10.1016/j.agwat.2009.03.016>.
- [21] Yadollahpour A., Rashidi S., Rezaee Z. & Jailifar M. (2014). Magnetic water treatment in environmental management. A review of the recent advances and future perspective. *Current World Environments*, 9(3), 1008–1016. Doi: <http://dx.doi.org/10.12944/CWE.9.3.56>.
- [22] Szatyłowicz E. & Skoczko I. (2015). Możliwości wykorzystania pola magnetycznego w inżynierii środowiska. In: I. Skoczko, J. Piekutina, E. Olszańska, E. Markiewicz (red.), *Inżynieria Środowiska – Młodym Okiem. Tom 13: Ekoinżynieria*. Białystok: Oficyna Wydawnicza Politechniki Białostockiej, 331–354. <https://www.researchgate.net/publication/313293278>.
- [23] Klassen V.I. (1973). *Voda i magnit*. Moskva: Nauka [Классен В.И. (1973). *Вода и магнит*. Москва: Наука].
- [24] Kartashev V.T., Pogosbekyan Yu.M. & Yefremov A.L. (1975). Magnitnaya i elektricheskaya obrabotka peschano-glinistykh smesey. *Liteynoye Proizvodstvo*, 11, 17–18 [Карташев В.Т., Погосбекян Ю.М., Ефремов А.Л. (1975). Магнитная и электрическая обработка песчано-глинистых смесей. *Литейное Производство*, 11, 17–18].
- [25] Tokarev A.I. & Belyakov A.I. (1973). Obrabotka svyazuyushchikh magnitnym polem i elektricheskim tokom. *Liteynoye Proizvodstvo*, 3, 30–31 [Токарев А.И., Беляков А.И. (1973). Обработка связующих магнитным полем и электрическим током. *Литейное Производство*, 3, 30–31].
- [26] Bobrovskikh I.Ye., Yanush Ye.I., Cherkas Z.A. & Nenakhov N.A. (1975). Issledovaniye vliyaniya skorosti prokhozhdeniya suspenzii cherez namagnichivayushchiy apparat na fiziko-mekhanicheskiye svoystva formovochnoy smesi. In: P.V. Chernogorov, Yu.P. Vasin, *Progressivnyye metody izgotovleniya liteynykh form*. Chelyabinsk: Chelyabinskoye izdatel'stvo, 274–276 [Бобровских И.Е., Януш Е.И., Черкас З.А., Ненахов Н.А. (1975). Исследование влияния скорости прохождения суспензии через намагничивающий аппарат на физико-механические свойства формовочной смеси. В книге: Черногоров П.В., Васин Ю.П., *Прогрессивные методы изготовления литейных форм*, 274–276. Челябинск: Челябинское издательство].

- [27] Kochkina T., Chumakova A., Vedenina T., Uvarova V. & Finogenova L. (1970). Omagnichennaya voda kak faktor povysheniya prochnosti peschano-tsementnykh formovochnykh smesey. In: A. Chumakova, *Progressivnyye metody i protsessy v liteynom proizvodstve*. Volgograd: Izdatel'stvo Nizhnevolzhskogo TSBTI, 22–23 [Кочкина Т., Чумакова А., Веденина Т., Уварова В., Финогонова Л. (1970). Омагниченная вода как фактор повышения прочности песчано-цементных формовочных смесей. В книге: А. Чумакова, *Прогрессивные методы и процессы в литейном производстве*. Волгоград: Издательство Нижневолжского ЦБТИ, 22–23].
- [28] Dan L.A., Trofimova L.A., Shepilov V.A. & Dan Ye.L. (2012). Povysheniye prochnostnykh svoystv syrykh peschano-glinistykh formovochnykh smesey putem elektromagnitnoy obrabotki vody. *Vestnik PGTU*, 24, 143–147 [Дан Л.А., Трофимова Л.А., Шепилов В.А., Дан Е.Л. (2012). Повышение прочностных свойств сырых песчано-глинистых формовочных смесей путем электромагнитной обработки воды. *Вестник ПГТУ*, 24, 143–147].
- [29] Ochkov V.F. (2011). Voda i magnit. Vodoochistka, Vodopodgotovka, Vodосnabzheniye, 10, 36–48 [Очков В.Ф. (2011). Вода и магнит. *Водоочистка, Водоподготовка, Водоснабжение*, 10, 36–48. Doi: <http://twf.mpei.ac.ru/ochkov/MO/BBB.html> [accessed: 21.03.2023].
- [30] Bernal D.J. & Fowler R.H. (1933). A theory of water and ionic solution, with particular reference to hydrogen and hydroxyl ions. *The Journal of Chemical Physics*, 1, 515–548.
- [31] Ramsay W. & Shields J. (1893). The molecular complexity of liquids. *Journal of the Chemical Society*, 63, 1089–1109.
- [32] Sutherland W. (1900). The molecular construction of water. *Philosophical Magazine*, 50, 460–489.
- [33] Stewart G.W. (1930). X-Ray Diffraction in liquids. *Reviews of Modern Physics*, 2, 116–122.
- [34] Mecke R. (1933). Des Rotationsschwingungsspektrum des Wasserdampfes. *Zeitschrift für Physik*, 81, 313–331.
- [35] Yel'yashevich M.A. (1938). Vrashchatel'no-kolebatel'naya energiya mnogoatomnykh molekul. *Trudy Gosudarstvennogo Opticheskogo Instituta*, 12(106), 39–40 [Ельяшевич М.А. (1938). Вращательно-колебательная энергия многоатомных молекул. *Труды Государственного Оптического Института*, 12(106), 39–40].
- [36] Magat M. (1936). Recherches sur le spectre Raman et la constitution de l'eau liquid. *Annales de Physique*, 11(6), 108–193.
- [37] Bjerrum N. (1952). Structure and Properties of Ice. *Science*, 115, 385–390.
- [38] Pauling L. (1935). The structure and entropy of ice and other crystal with some randomness of atomic arrangement. *Journal of the American Chemical Society*, 57, 2680–2684.
- [39] Lennard-Jones J. & Pople J.A. (1950). The molecular orbital theory of chemical valency. IV. The significance of equivalent orbitals. *Proceedings of the Royal Society A*, 202, 166–180. Doi: <https://doi.org/https://doi.org/10.1098/rspa.1950.0092>.
- [40] Pang X. (2006). The conductivity properties of protons in ice and mechanism of magnetization of liquid water. *The European Physical Journal B*, 49, 5–23. Doi: <https://doi.org/10.1140/epjb/e2006-00020-6>.
- [41] Pang X.F. & Deng B. (2008). Investigation of changes in properties of water under the action of a magnetic field. *Science China Physics, Mechanics & Astronomy*, 51, 1621–1632. Doi: <https://doi.org/10.1007/s11433-008-0182-7>.
- [42] Amiri M.C. & Dadkhah A.A. (2006). On reduction in the surface tension of water due to magnetic treatment. *Colloids and Surfaces A: Physicochemical and Engineering Aspects*, 278(1–3), 252–255. <https://doi:10.1016/j.colsurfa.2005.12.046>.
- [43] Premkaisorn P. & Wasupongpun W. (2020). Magnetic field effect on physicochemical properties of water. *RMUTP Research Journal*, 14(2), 1–17.
- [44] Deng B. & Pang X. (2007). Variations of optic properties of water under action of static magnetic field. *Chinese Science Bulletin*, 52, 3179–3182. Doi: <https://doi:10.1007/s11434-007-0430-7>.
- [45] Krestov G.A. (1984). *Termodinamika ionnykh protsessov v rastvorakh*. Leningrad: Nauka [Крестов Г.А. (1984). *Термодинамика ионных процессов в растворах*. Ленинград: Наука].
- [46] Martynova O.I., Gusev B.T. & Leont'yev Ye.A. (1969). O mekhanizme vliyaniya magnitnogo polya na vodnyye rastvory soley. *Uspekhi Fizicheskikh Nauk*, 98, 25–31 [Мартынова О.И., Гусев Б.Т., Леонтьев Е.А. (1969). О механизме влияния магнитного поля на водные растворы солей. *Успехи Физических Наук*, 98, 25–31].
- [47] Kronenberg K. (1985) Experimental evidence for the effects of magnetic fields on moving water. *IEEE Transactions on Magnetics (Institute of Electrical and Electronics Engineers)*, 21(5), 2059–2061.
- [48] Ignatov I.I. & Mosin O.V. (2013). Structural mathematical models describing water clusters. *Nanotechnology Research and Practice*, 3(11), 72–87. Doi: <https://doi.org/10.13187/ejnr.2014.3.141>.
- [49] Boldin A.N., Davydov N.I. & Zhukovskiy S.S. (2006). *Liteynyye formovochnyye materialy. Formovochnyye, sterzhnevyye smesi i pokrytiya. Spravochnik*. Moskva: Mashinostroyeniye [Болдин А.Н., Давыдов Н.И., Жуковский С.С. (2006). *Литейные формовочные материалы. Формовочные, стержневые смеси и покрытия. Справочник*. Москва: Машиностроение].
- [50] Gorbunov N.I. (1957). *Pochvennyye kolloidy*. Moskva: Izdatel'stvo Akademii Nauk SSSR [Горбунов Н.И. (1957). *Почвенные коллоиды*. Москва: Издательство Академии Наук СССР].
- [51] Tolstoy N.A. (1967). Zhestkiy elektricheskiy dipol'nyy moment kolloidnykh chastits. In: B.V. Deryagin (red.), *Issledovaniya v oblasti poverkhnostnykh sil*. Moskva: Nauka, 56–78 [Толстой Н.А. (1967). Жесткий электрический дипольный момент коллоидных частиц. В книге: Б.В. Дерягин (Ред.), *Исследования в области поверхностных сил*. Москва: Наука, 56–78].
- [52] Illarionov I.Ye. (2011). Teoreticheskiye osnovy formirovaniya fiziko-mekhanicheskikh svoystv peschano-glinistykh smesey. *Trudy Nizhegorodskogo Gosudarstvennogo Tekhnicheskogo Universiteta im. R.Ye. Alekseyeva*, 1(86), 233–241 [Илларионов И.Е. (2011). Теоретические основы формирования физико-механических свойств песчано-глинистых смесей. *Труды Нижегородского Государственного Технического Университета им. Р.Е. Алексеева*, 1(86), 233–242].
- [53] Vasin Yu.B., Semenchenko I.B., Bortnikov M.M., Gorlov V.V. & Vasina Z.M. (1969). *Progressivnaya tekhnologiya liteynogo proizvodstva*. Gor'kiy: Gor'kovskoye knizhnoye izdatel'stvo [Васин Ю.Б., Семенченко И.Б., Бортников М.М., Горлов В.В., Васина З.М. (1969). *Прогрессивная технология литейного производства*. Горький: Горьковское книжное издательство].